

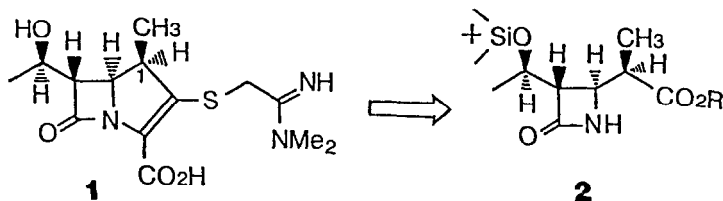
A NOVEL, DOUBLE-ASYMMETRIC ALDOL APPROACH TO THE SYNTHESIS OF A 1 β -METHYL
 CARBAPENEM ANTIBIOTIC PRECURSOR

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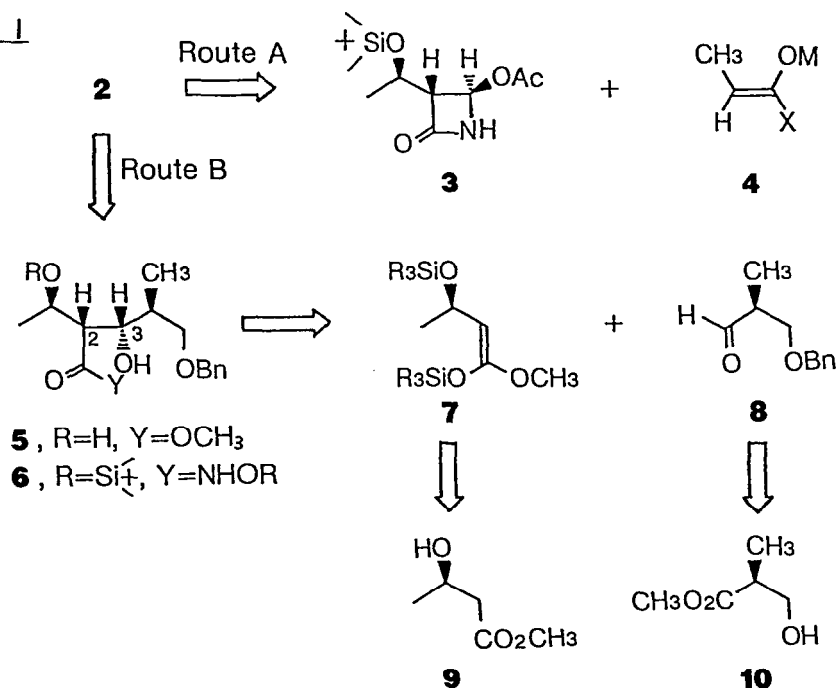
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SUMMARY: A novel synthetic approach to the 1 β -methyl carbapenem key precursor is described which involves the chelation-controlled double-asymmetric aldol reaction as the key step.

Since the discovery of the enhanced chemical and metabolic stability of 1 β -methyl carbapenem antibiotics such as **1**,² considerable effort has been devoted to the stereocontrolled synthesis of the optically active key precursor **2**.³ The most popular and successful entries to **2** have so far relied on the aldol-type reaction of the (+)-acetoxymethyl azetidinone **3**⁴ with properly designed chiral^{3b,c} or achiral^{3d-f} metal enolates **4** (Route A in Scheme I). Retrosynthetic analysis suggested an entirely different approach (Route B) which



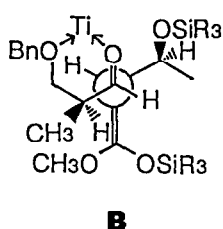
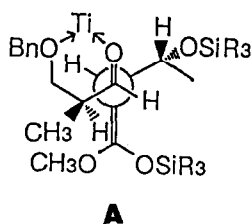
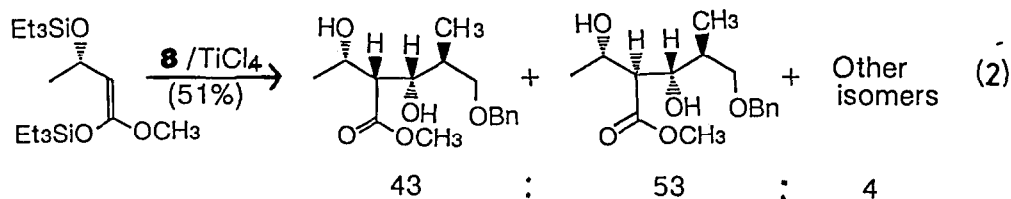
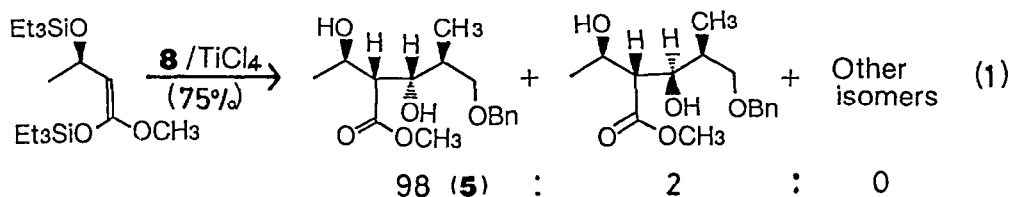
Scheme I



would incorporate the connection of the chiral "left-half" **7** with the chiral "right-half" **8** via an aldol reaction followed by the well-established $N-C_4$ cyclization of hydroxamate **6**.⁵ We now report the successful realization of this novel approach starting with the two optically active β -hydroxy esters **9** and **10**, both commercially available in quantity. The key feature is the remarkable high efficiency of consonant double stereodifferentiation in the double-asymmetric aldol process.⁶

Stereochemical analysis of the asymmetric aldol process led us to employ chelation-controlled conditions.^{7,8} Thus, we carried out reactions of the silyl ketene acetal **7** (R=Me or Et)⁹ with the β -benzyloxy aldehyde **8** derived from **10**¹⁰ in dichloromethane at $-70 \sim -40$ °C under Lewis acid conditions [$TiCl_4$, (*i*-PrO) $TiCl_3$, or $SnCl_4$]. We found that the aldol reaction was best achieved by the specific combination of **7** (R=Et) with $TiCl_4$ to afford 75% yield¹¹ of the desired aldol **5** in a remarkably high diastereomeric purity ($\geq 97\%$) as shown by eq 1.^{12,13} The relative stereochemistry of **5**, while eventually confirmed by its conversion to **2**, could be deduced from 500 MHz 1H NMR comparison of its acetonide with other diastereomeric acetonides separately prepared.^{8,14} Thus the crucial aldol reaction established the four contiguous chiral centers required.

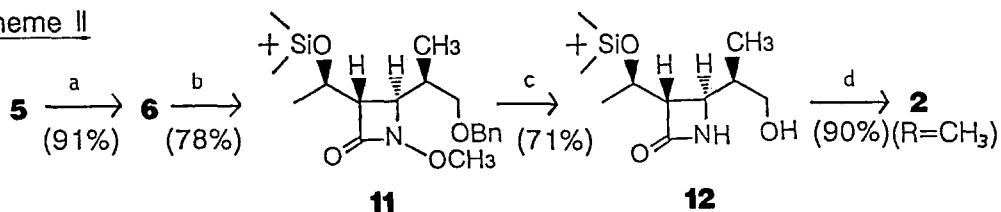
As expected, a similar reaction of the antipode of **7** with **8** resulted in a dramatic decrease in diastereofacial selectivity (eq 2).¹⁵ Thus, the chelation-controlled aldol reaction of the $7(R)/8(S)$ pair is clearly demonstrated to show the favorable double stereodifferentiation in which the inherent diastereoface preference of the two reactants may reinforce one another, as visualized the transition state **A**, compared with **B** for the $7(S)/8(S)$ pair.



The transformation of the aldol **5** to the key precursor **2** is straightforward (Scheme II). Thus, aldol **5** was converted, after selective silylation, to hydroxamate **6** by the modification

of Weinreb's method.^{16,17} Cyclization of **6** was best achieved by the mesylation/hydroxaminolysis sequence⁵ to give the β -lactam **11** with >95% diastereomeric purity (HPLC).¹⁸ The β -lactam **11** was then subjected to the Birch reduction to give the azetidinone **12**.¹⁹ Finally, oxidation of **12** followed by treatment with diazomethane furnished the desired key precursor **2** (R=CH₃) in essentially 100% ee. Its physical and spectral data were identical with those reported [mp 119–120 °, $[\alpha]_D^{25}$ -21.1° (c 0.60, CH₂Cl₂); lit.² mp 120–121 °C, $[\alpha]_D^{25}$ -21.0° (c 2.09, CH₂Cl₂)].

Scheme II



- (a) (i) *t*-BuMe₂SiCl, imidazole, DMF, 25°C; (ii) AlMe₃/MeONH₂HCl, toluene, 25°C.
 (b) (i) MeSO₂Cl, pyridine, 25°C; (ii) K₂CO₃, MeOH, 25°C. (c) Li, EtNH₂-*t*-BuOH-THF (3:1:1), -40°C. (d) (i) CrO₃, pyridine, 25°C; (ii) CH₂N₂, Et₂O, 25°C

In summary, we have developed a novel, efficient approach to the 1 β -methyl carbapenem precursor employing the chelation-controlled double-asymmetric aldol reaction as the key step. The easy availability of the optically active starting materials, coupled with the high efficacy of stereocontrol, places the present approach in a unique position for the practical synthesis of 1 β -methyl carbapenem antibiotics. Further improvement of this approach as well as mechanistic studies on this interesting aldol process are under way.

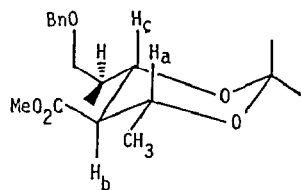
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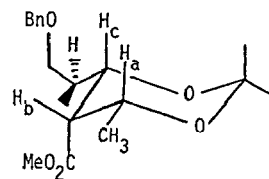
see: ref 1f.

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- (7) Review on chelation- vs. non-chelation control in aldol and related reactions. Reetz, M, T. *Angew. Chem. Int. Ed. Engl.* **1984**, 23, 556.
- (8) We found that the non-chelation-controlled reaction of the dilithium enolate of **9** with **8** provided a 1:1:2:6 mixture of the aldol diastereomers (63% yield). The major diastereomer was not **5** but the 2-epimer of **5** based on the HPLC and NMR of its acetonide (cf. ref 14).
- (9) The silyl ketene acetals **7** (R=Me and Et) were prepared by silylation of the dilithium (Z)-enolate generated from **9** (LDA, THF) with Me₃SiCl and Et₃SiCl, respectively, in ca. 75% distilled yields. A similar silylation with t-BuMe₂SiCl was unsuccessful.
- (10) The procedure of A. I. Meyers, et al. [*J. Am. Chem. Soc.* **1983**, 105, 5015] was adopted.
- (11) The aldol yields varied critically in the range of 55-75% with slight changes in timing and rate of addition of **7** to a mixture of **8** and TiCl₄. A major byproduct was methyl crotonate, an elimination product of **7**, and unreacted aldehyde was recovered unchanged.
- (12) The stereopurity was determined by HPLC [Zorbax SIL, hexane/AcoEt (4:1)] of the product mixture and confirmed by capillary GLC (Ulbon HR-20M, 50 m) of its acetonide.
- (13) Other combinations such as 7(R=Me)/TiCl₄, 7(R=Me)/(i-PrO)TiCl₃, and 7(R=Et)/(i-PrO)TiCl₃ provided much lower yields (15%, 32%, and 10%, respectively), although the stereopurities of the aldol were comparably high (≥95%). The combination of 7(R=Et)/SnCl₄ afforded 30% yield of **5** in only 79% diastereomeric purity.
- (14) The most informative are the coupling constants for the diastereomeric acetonides as exemplified below. For similar stereochemical assignments of structurally related



(acetonide of **5**)

$$J_{ab}=10.5 \text{ Hz}, J_{bc}=10.5 \text{ Hz}$$



(acetonide of the 2-epimer of **5**)

$$J_{ab}=3.0 \text{ Hz}, J_{bc}=2.7 \text{ Hz}$$

acetonides, see: Thaisrivongs, S.; Seebach, D. *J. Am. Chem. Soc.* **1983**, 105, 7407. The stereochemical assignment of each diastereomer will be fully discussed in a full paper.

- (15) The stereoisomeric ratio and the relative stereochemistry were determined in the same ways as described above.
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- (17) The conventional hydrolysis/hydroxaminolysis sequence⁵ provided only 50% yield of **6**.
- (18) Direct cyclization via the Mitsunobu reaction (dimethyl azodicarboxylate/Ph₃P) afforded only 38% yield of **11**.
- (19) Under the conditions described in Scheme II, 29% of **11** was recovered. Particularly noteworthy is that the presence of t-BuOH is essential for deprotecting both the N-methoxy and benzyl groups, otherwise only the N-demethoxylation product was obtained quantitatively.

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